

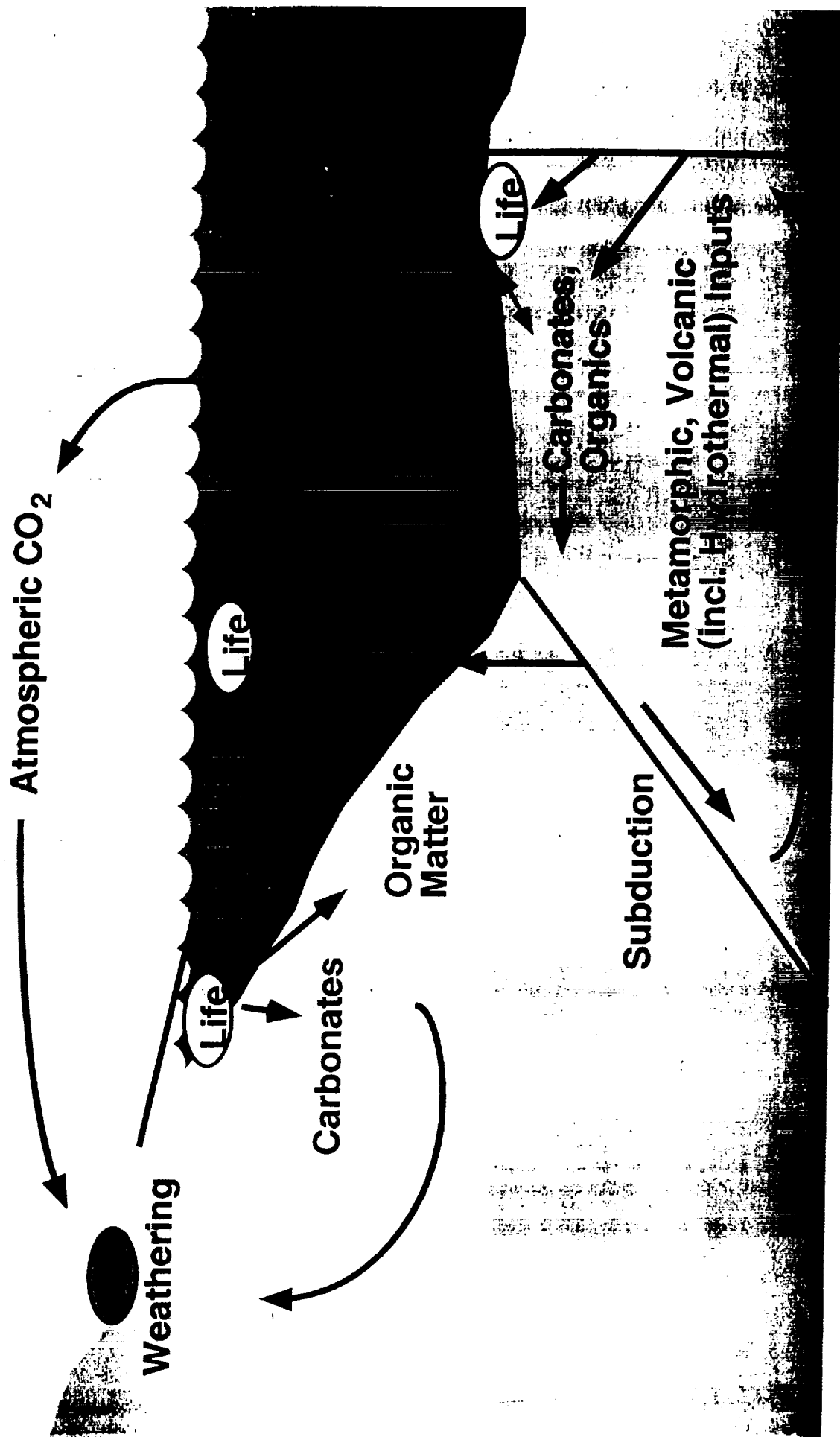
Figure 17.1. Oxygen cycle on Earth today, showing processes that are significant in producing or destroying oxygen. Chemical reactions involving oxygen are summarized; the actual chemistry involves many more steps than the equations on the figure show. Based on Cloud (1986).

Sedimentary Terminal Electron Acceptors

"Bounce per Ounce"

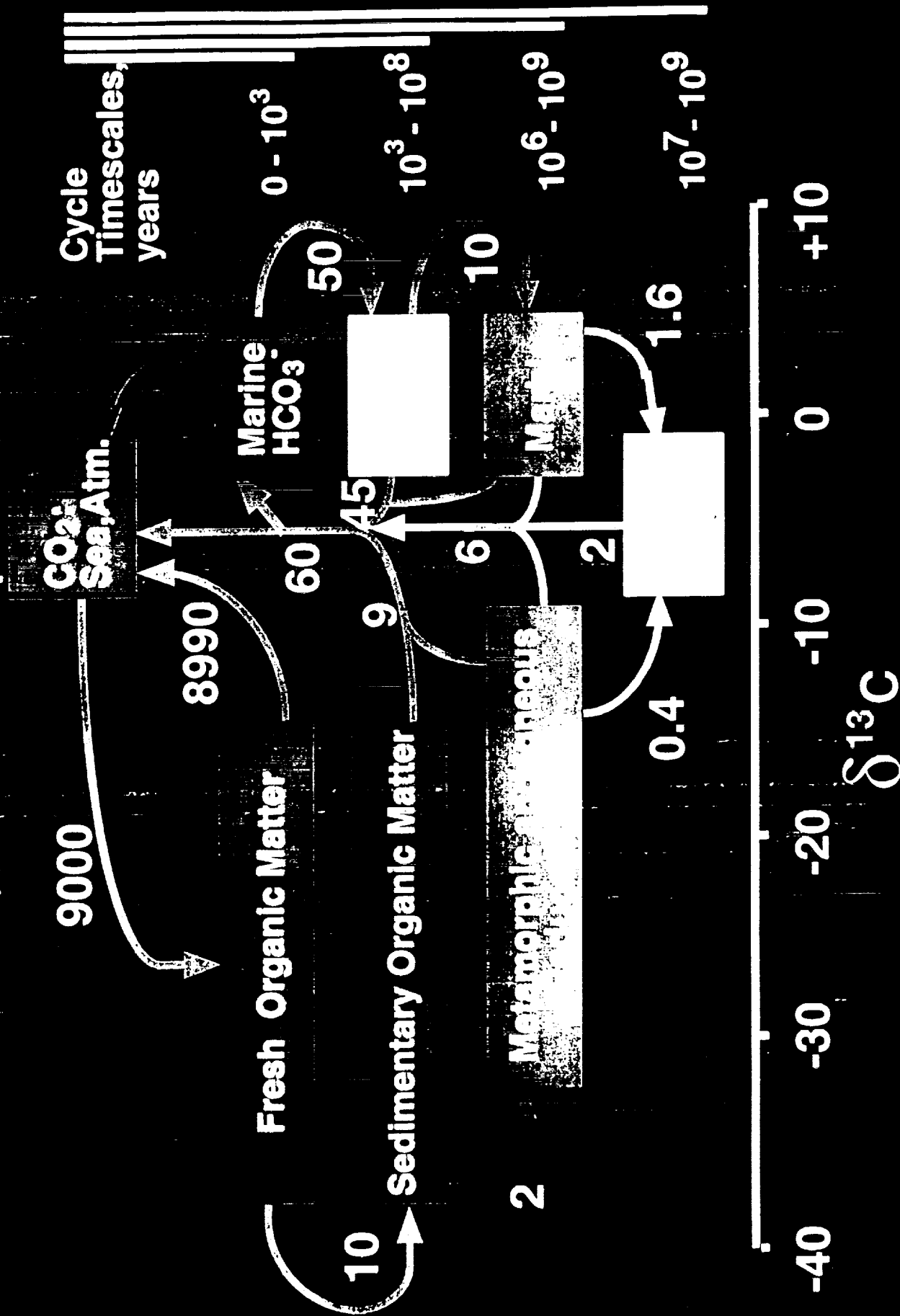
Reaction	Stoichiometry	ΔG° (kJ mol ⁻¹ of CH ₂ O)
Oxic Respiration	$\text{CH}_2\text{O} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$	-475
Denitrification	$5\text{CH}_2\text{O} + 4\text{NO}_3^- \rightarrow 2\text{N}_2 + 4\text{HCO}_3^- + \text{CO}_2 + 3\text{H}_2\text{O}$	-448
Mn-oxide Reduction	$\text{CH}_2\text{O} + 3\text{CO}_2 + \text{H}_2\text{O} + 2\text{MnO}_2 \rightarrow 2\text{Mn}^{2+} + 4\text{HCO}_3^-$	-349
Fe-oxide Reduction	$\text{CH}_2\text{O} + 7\text{CO}_2 + 4\text{Fe}(\text{OH})_3 \rightarrow 4\text{Fe}^{2+} + 8\text{HCO}_3^- + 3\text{H}_2\text{O}$	-114
Sulfate Reduction	$\text{CH}_2\text{O} + \text{SO}_4^{2-} \rightarrow \text{H}_2\text{S} + 2\text{HCO}_3^-$	-77
Methane Production	$2\text{CH}_2\text{O} + 2\text{H}_2\text{O} \rightarrow 2\text{CO}_2 + 4\text{H}_2$, and $4\text{H}_2 + \text{CO}_2 \rightarrow \text{CH}_4 + 2\text{H}_2\text{O}$	-58

Biogeochemical Carbon Cycle



Biogeochemical Carbon Cycle

Fluxes, $\times 10^{12}$ Moles per Year



Earth's Carbon Budget

Biosphere, Oceans and Atmosphere • 3.7×10^{18} moles

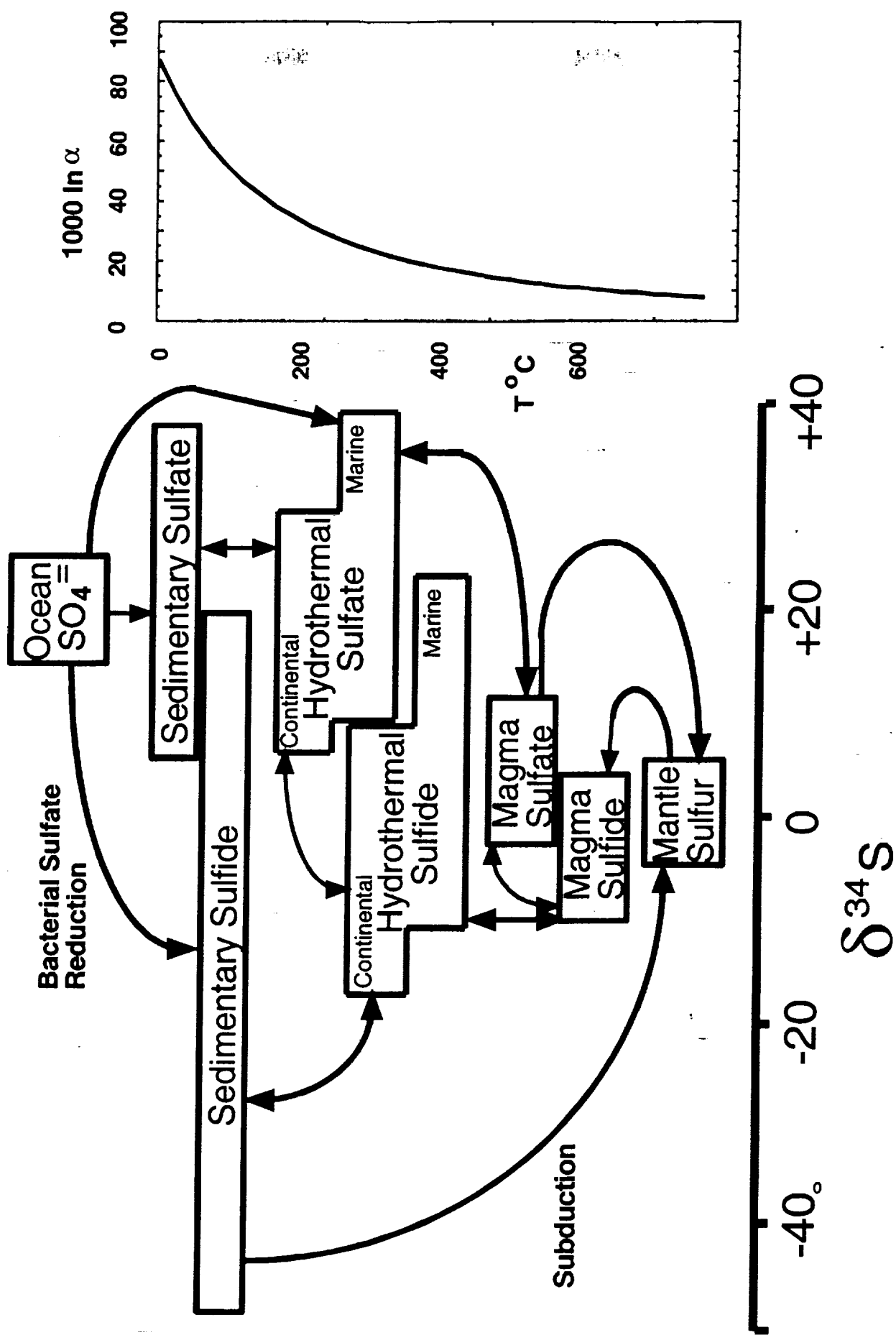
Crust

Organics
 1100×10^{18}
moles

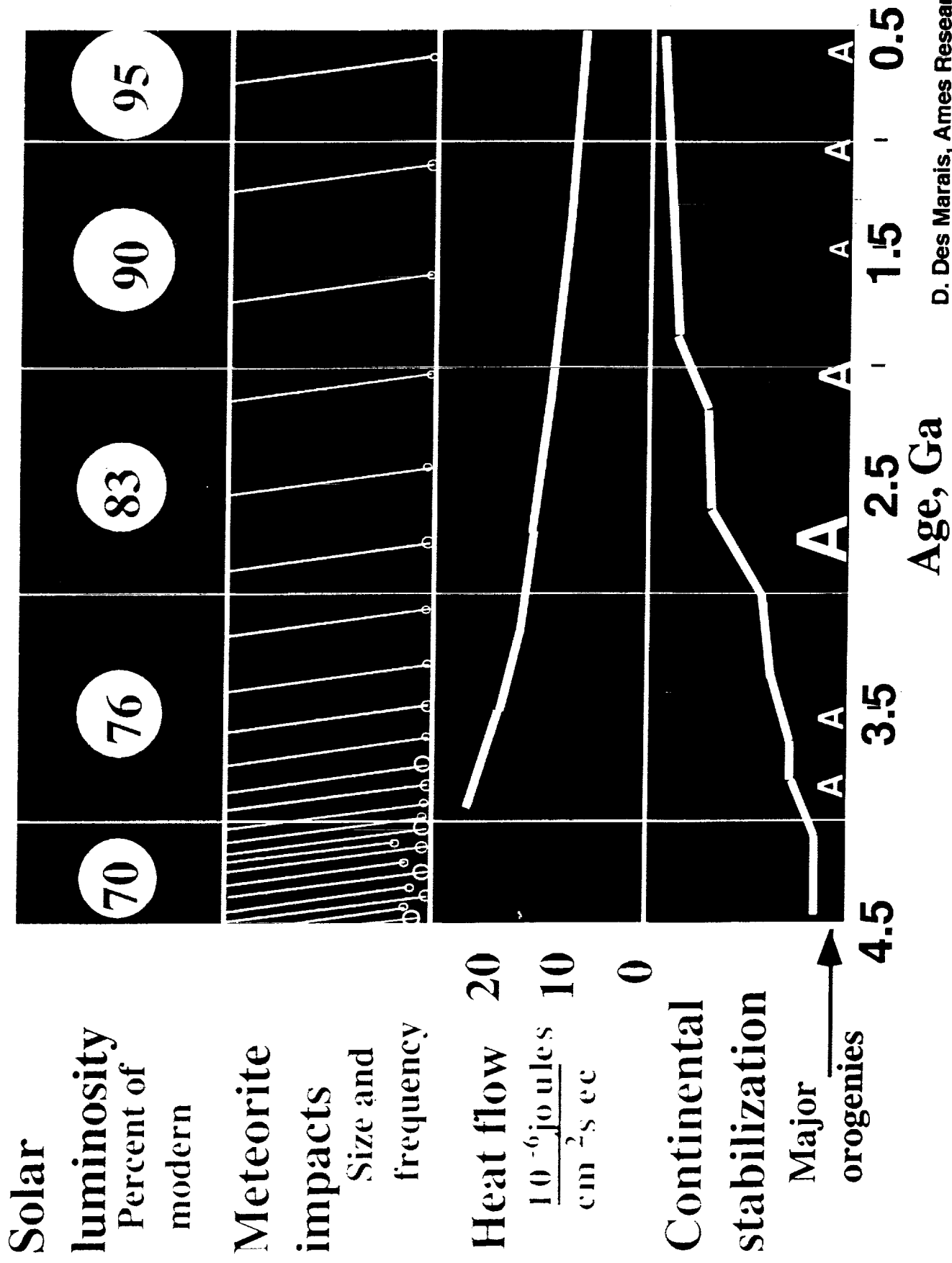
Carbonate
 5200×10^{18} moles

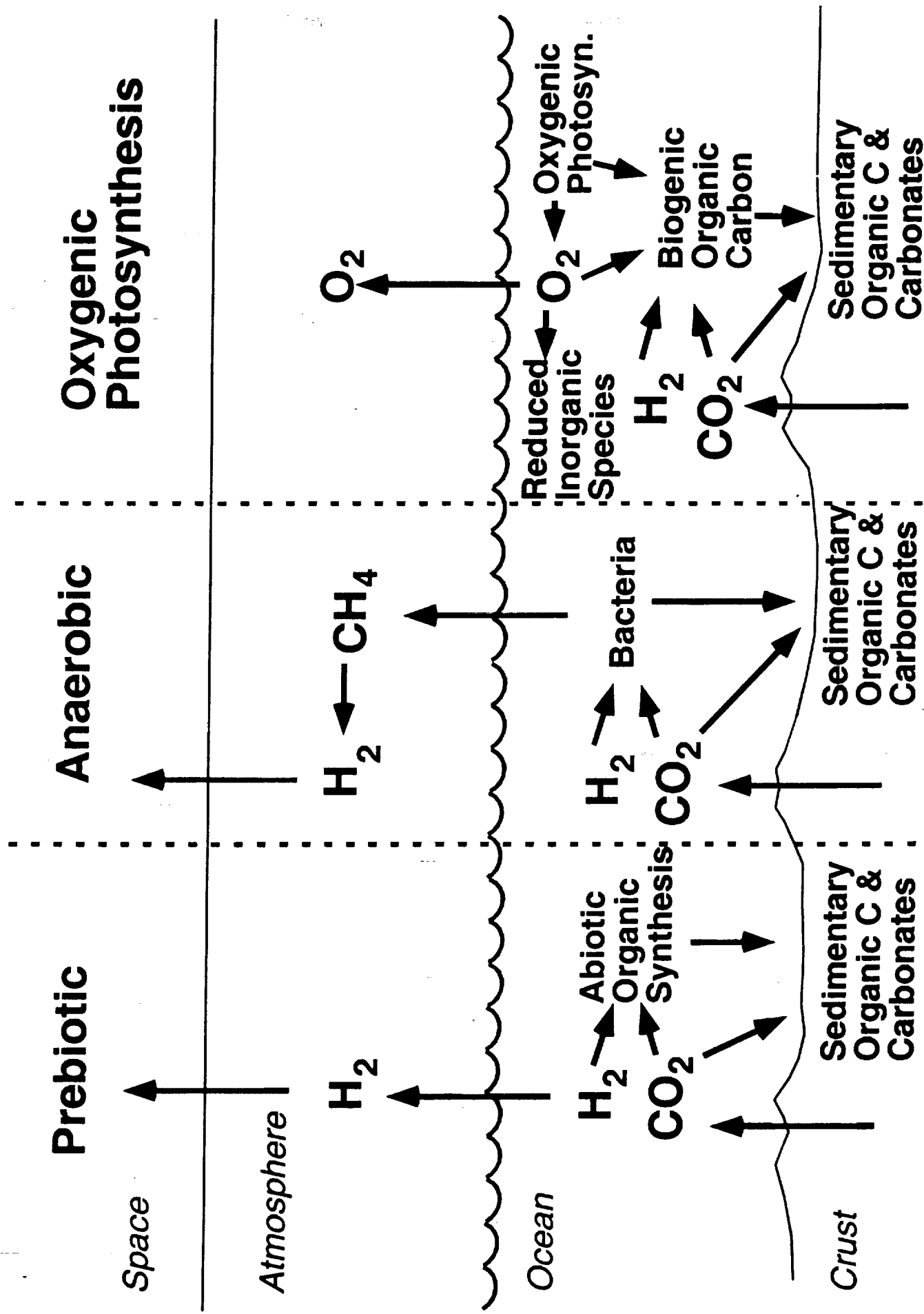
Mantle

$25,000 \times 10^{18}$ moles



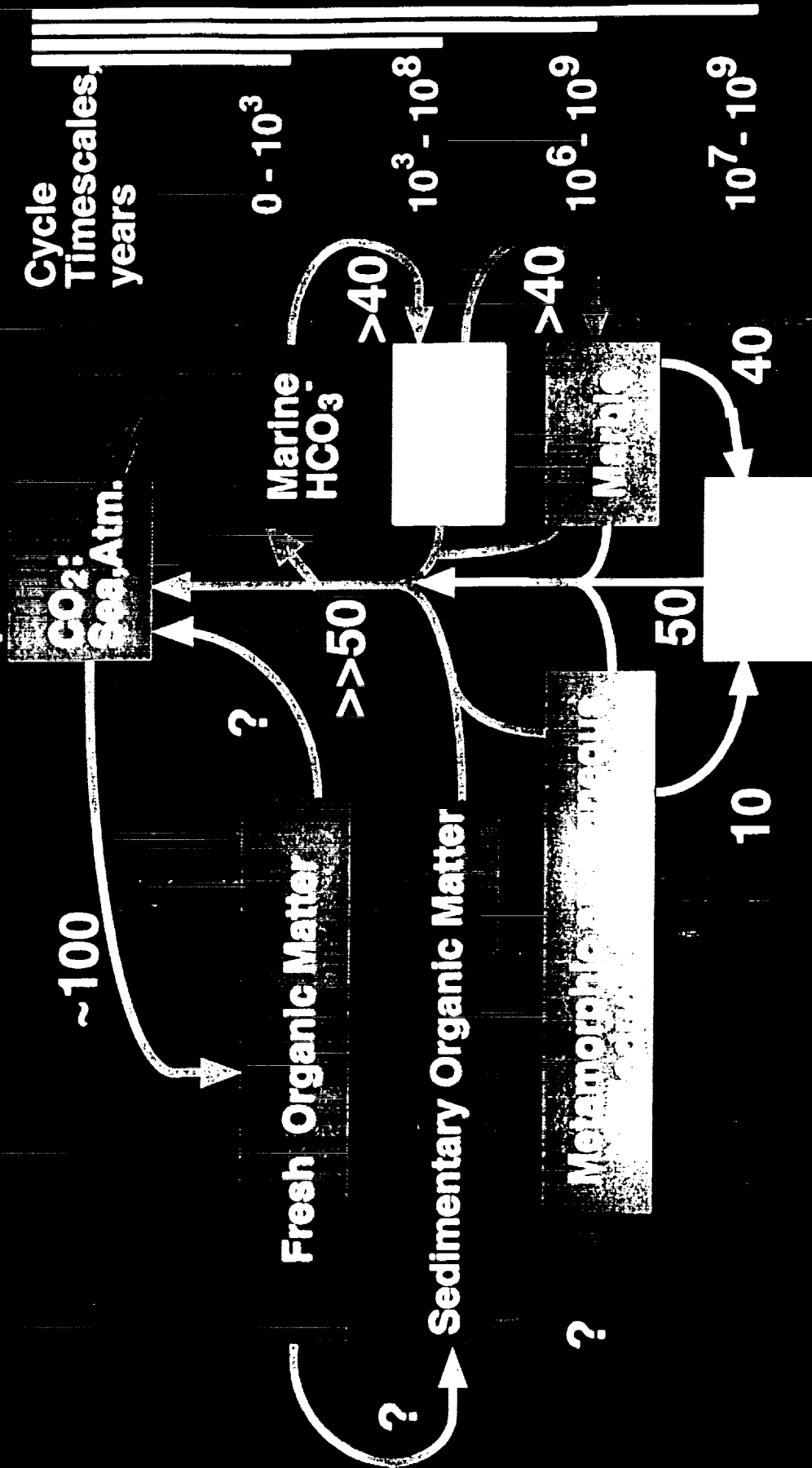
Evolution of Earth's Early Environment





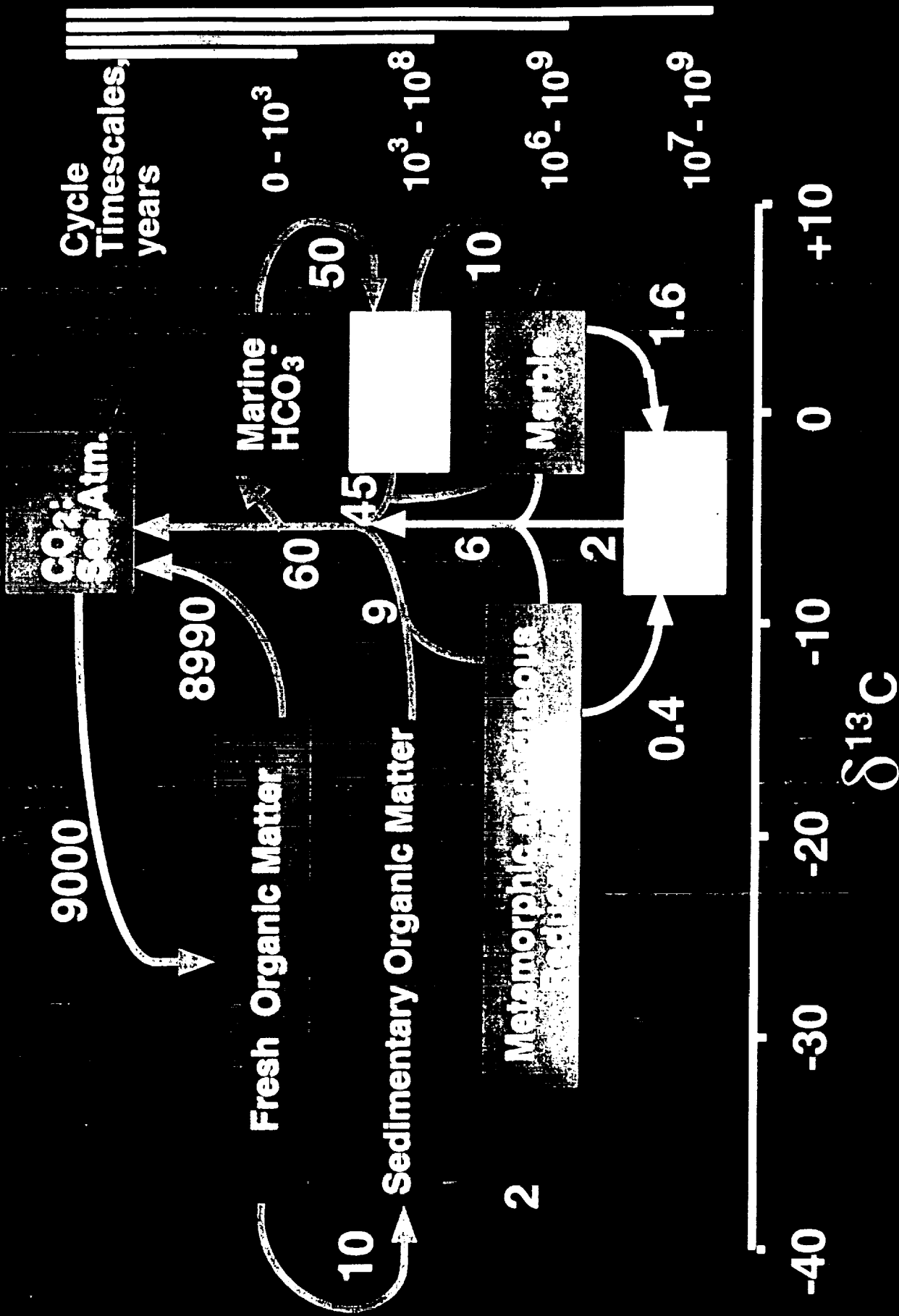
Ancient "Pre-O₂-photosynthesis" Carbon Cycle

Fluxes, $\times 10^{12}$ Moles per Year



Biogeochemical Carbon Cycle

Fluxes, $\times 10^{12}$ Moles per Year



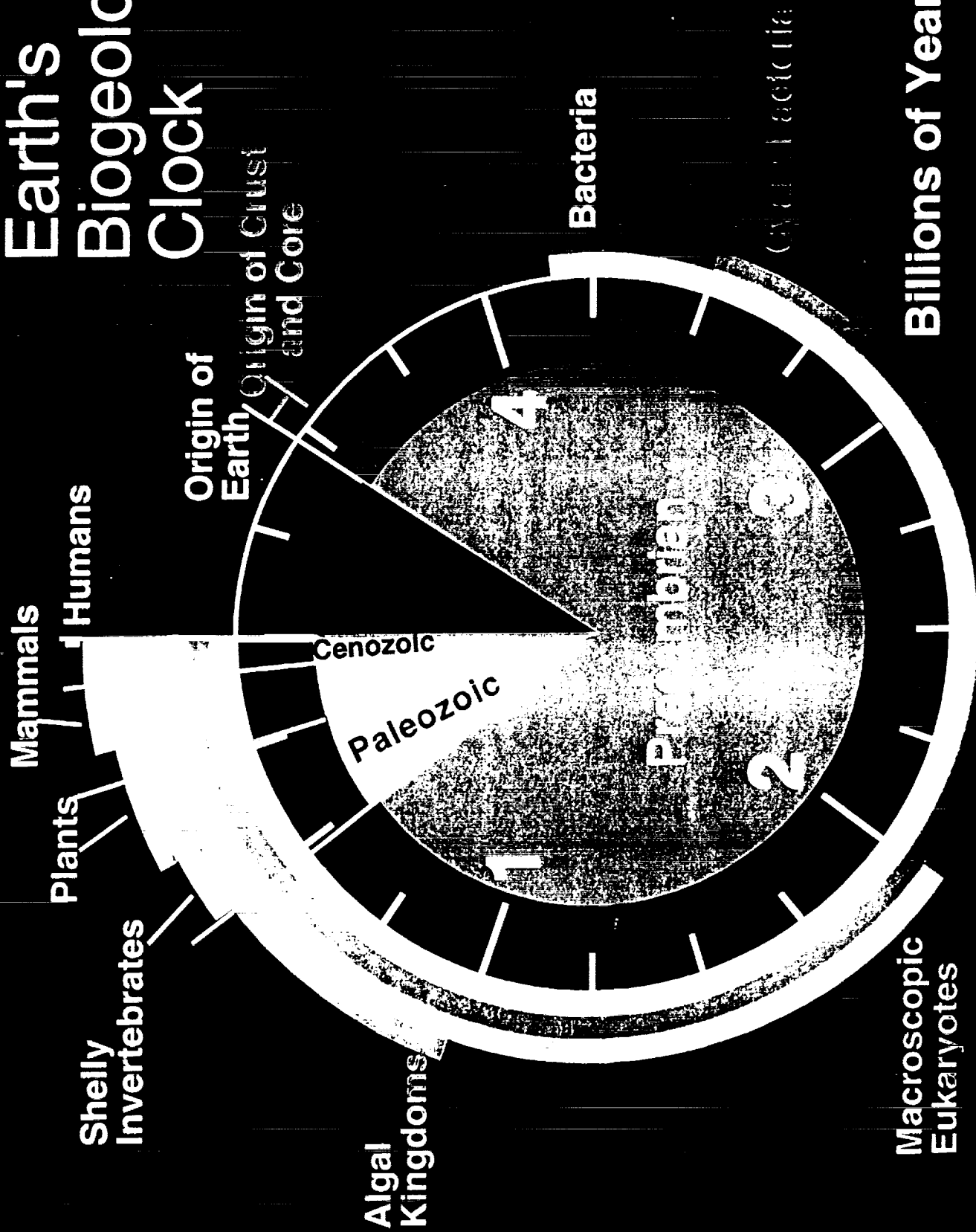
Oxygenic photosynthesis in the Archean (Part 1)

- Microfossils strongly resembling cyanobacteria: segmented filaments, coccoid clusters & sheaths having “correct” dimensions (Apex: 3.4 Ga, Fortescue: 2.7 Ga)
- Stromatolites & biofilms in lakes and shallow marine environments (not near thermal vents, etc.) consistent with O₂ photosynthetic communities (Fig Tree? & Warrawoona?: 3.4+ Ga; Pongola: 3.0 Ga, Fortescue: 2.7 Ga)
- Isotopically very light kerogens and normal carbonates consistent with a consortium of CO₂ autotrophs-methanogens-methanotrophs (Fortescue & Venterdorp: 2.7 Ga)

Oxygenic photosynthesis in the Archean (Part 2)

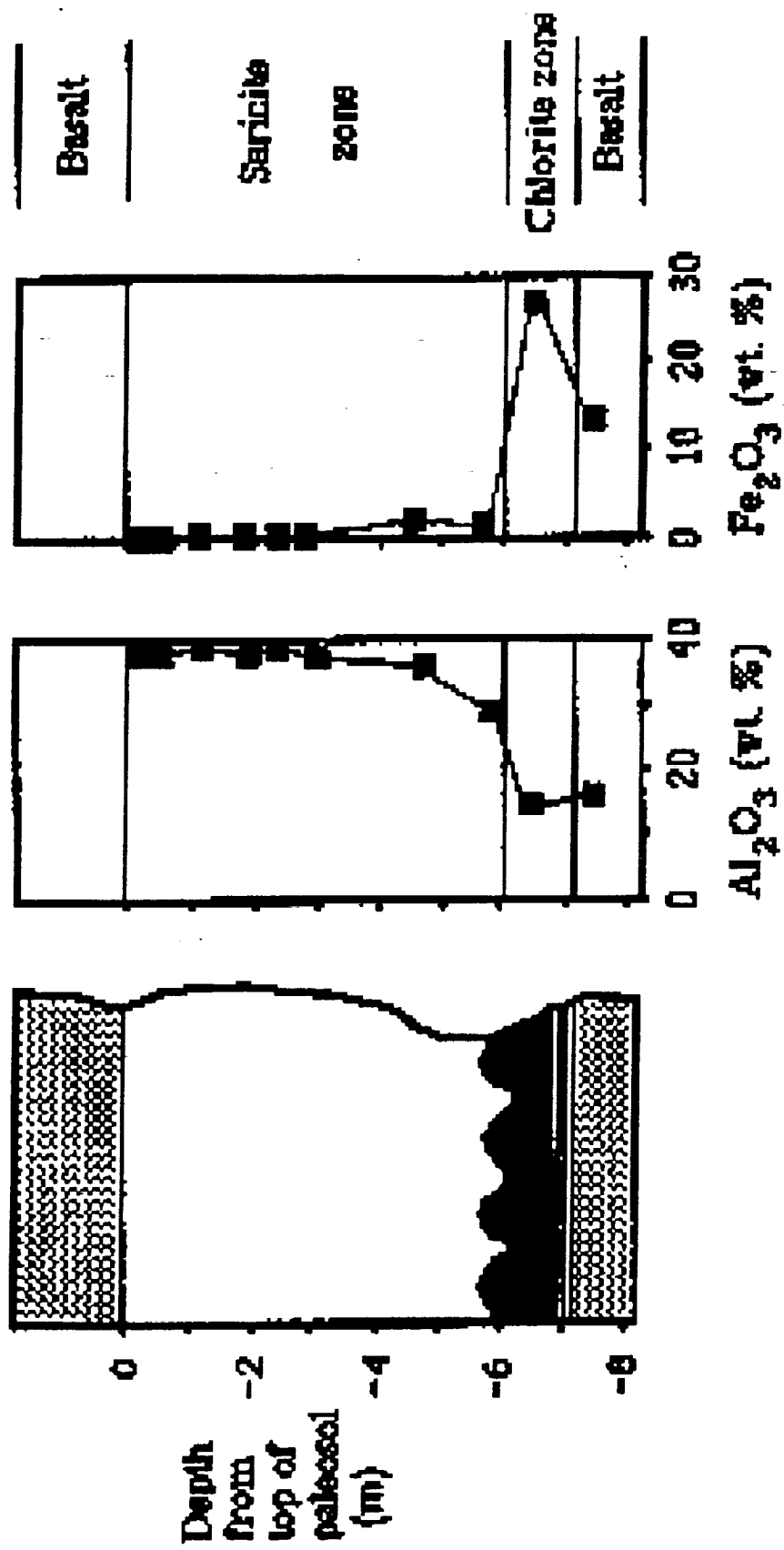
- Robust carbonate platform & reef formation closely analogous to modern examples: high-productivity communities (Transvaal: 2.5 Ga and younger)
- Oxidation of Fe^{2+} to Fe^{3+} in outer marine platform margins below the photic zone, with net deposition of Fe^{3+} (i.e., little or no local bio-organosynthesis)
- Organic concentrations in Archean cherts, carbonates and shales very similar to organic concentrations in Proterozoic and Phanerozoic lithologies
- Geochemical evidence for SRB at 2.7 Ga (McRae Sh), together with rRNA evidence for simultaneous divergence of SRB and cyanobacterial lineages

Earth's Biogeologic Clock

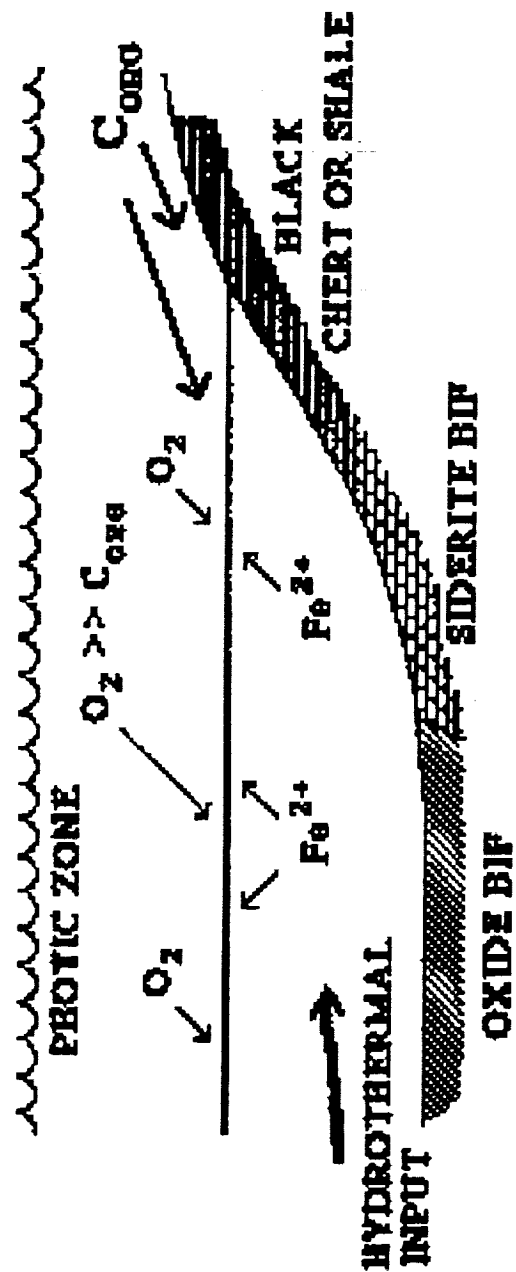


Billions of Years Ago

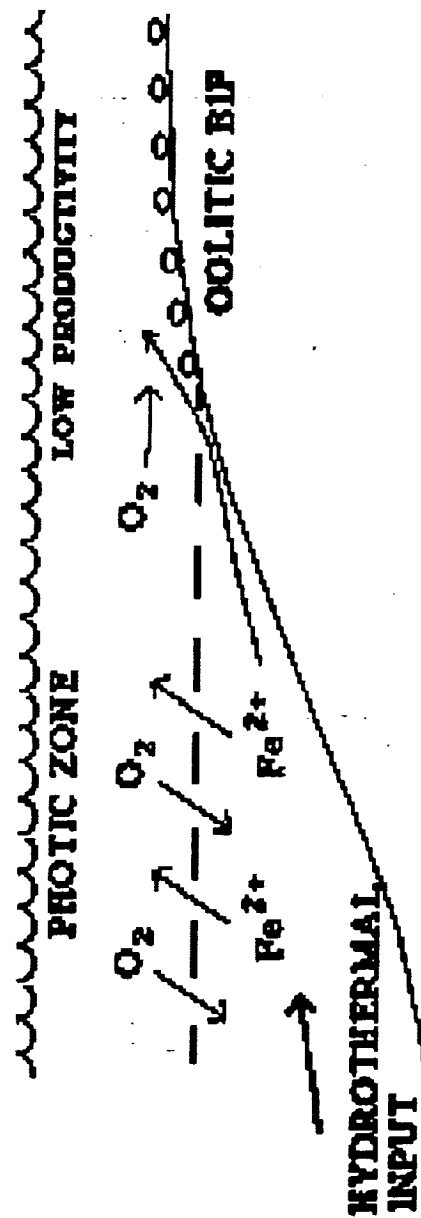
D. Des Marais
NASA, Ames Research Center



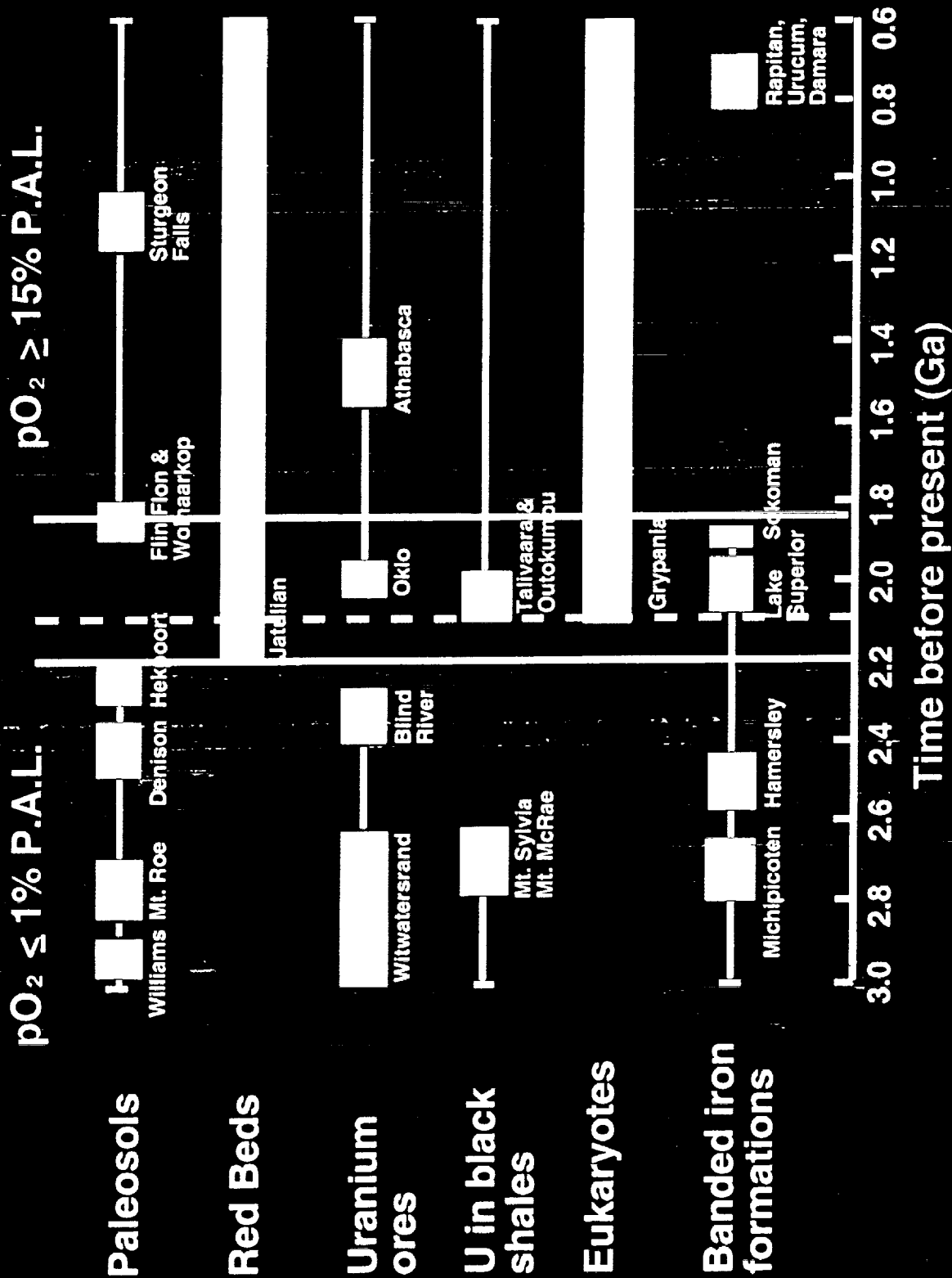
A

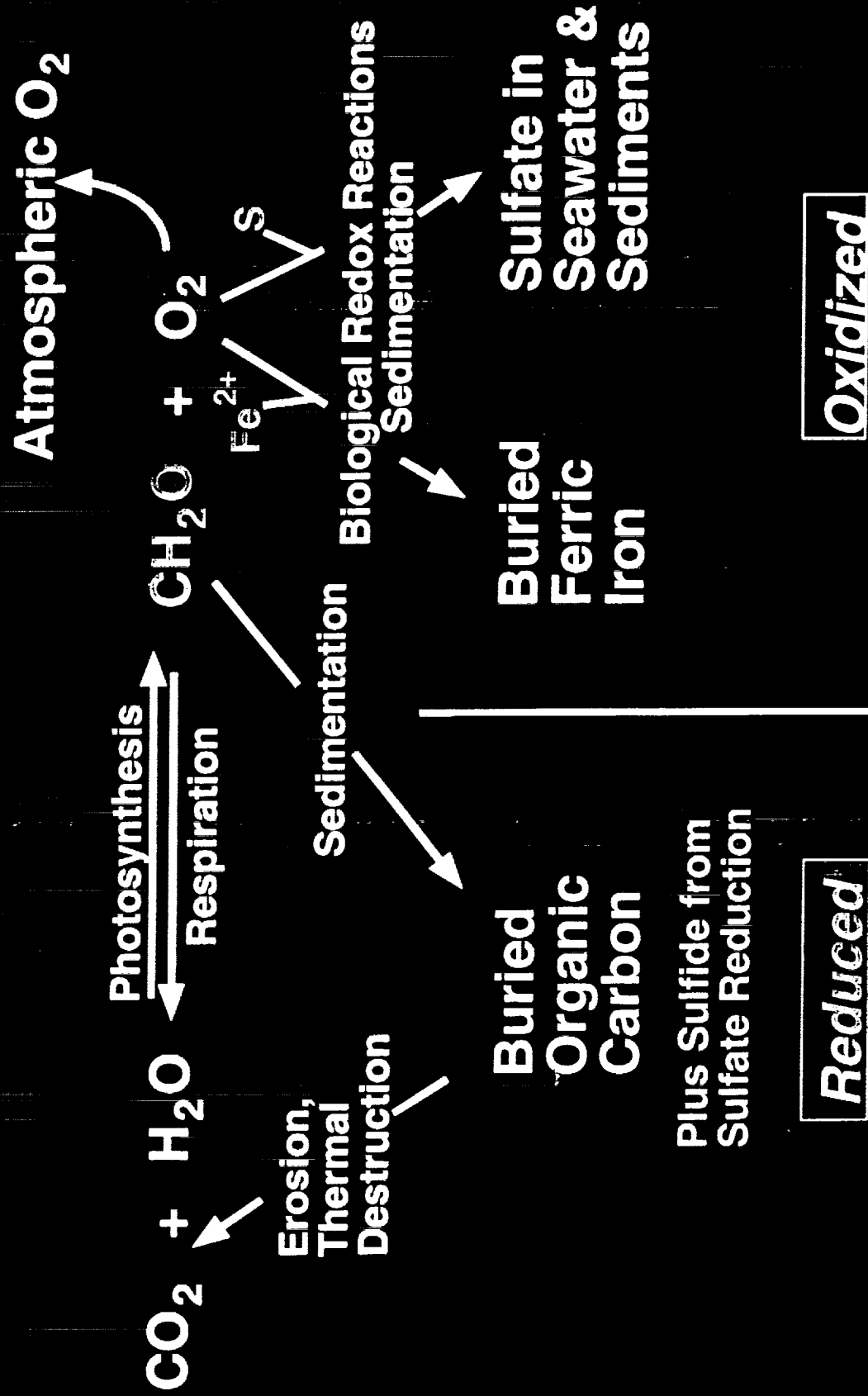


B

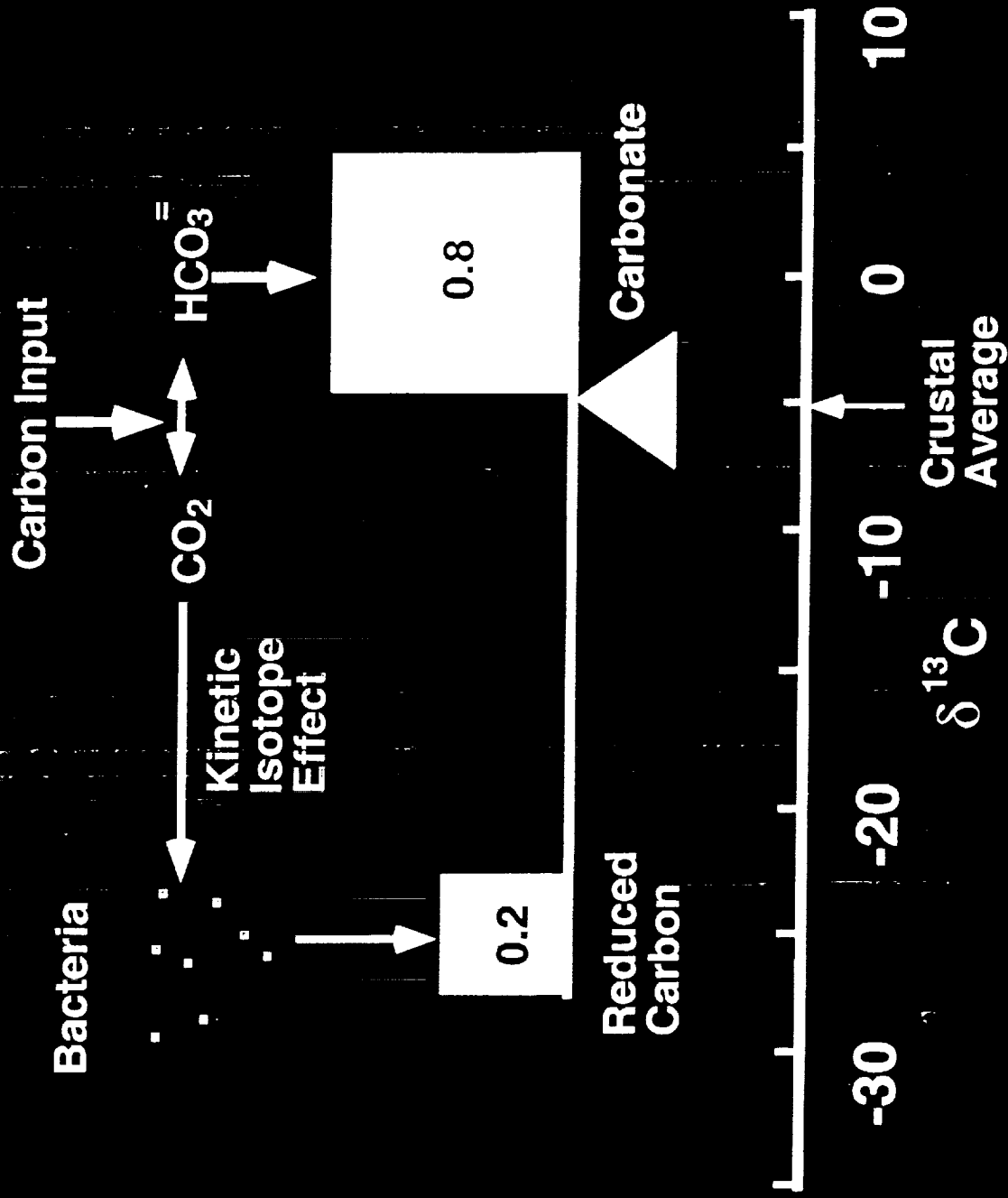


Evidence for the Oxidation of Earth's Environment

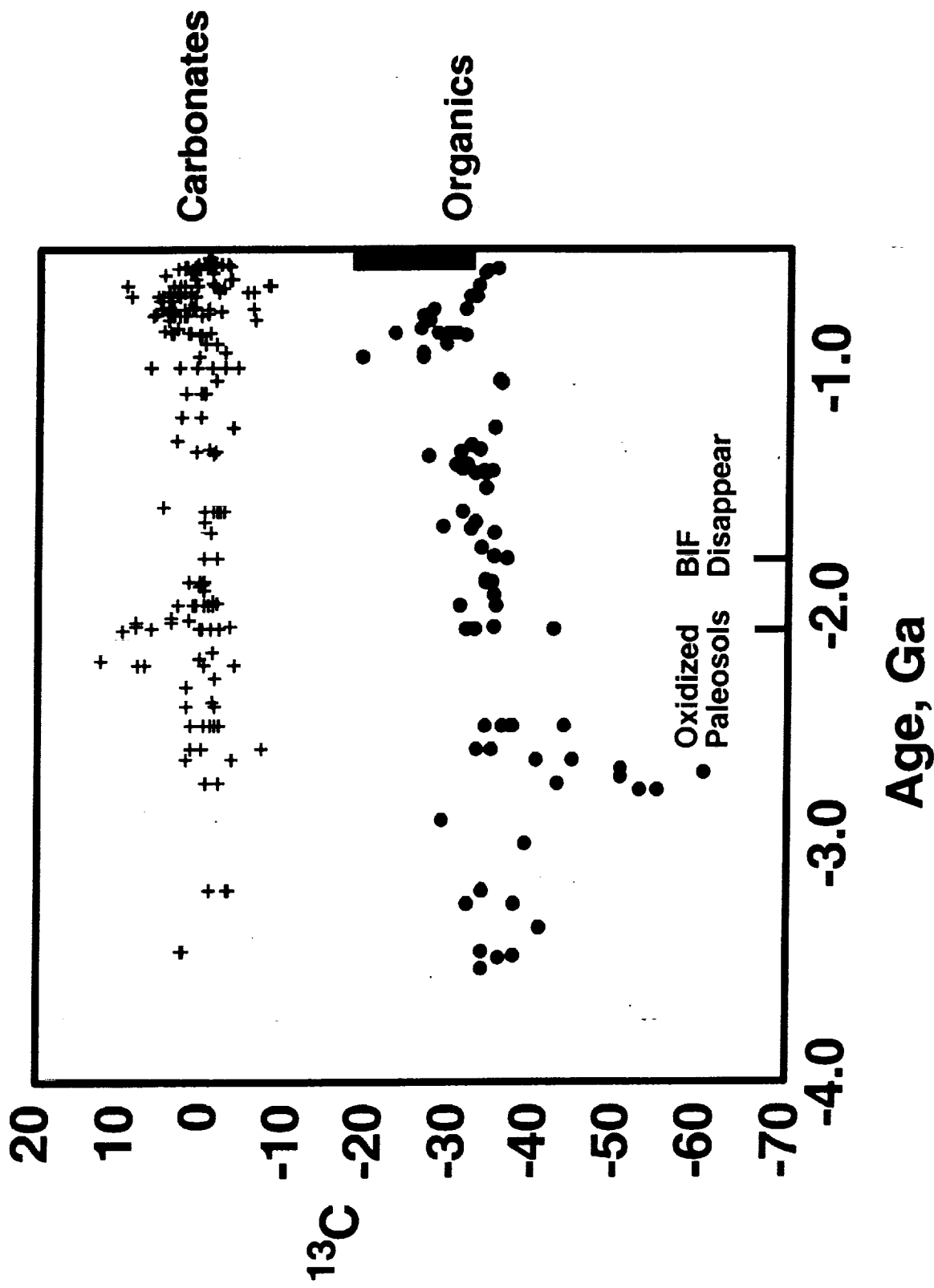


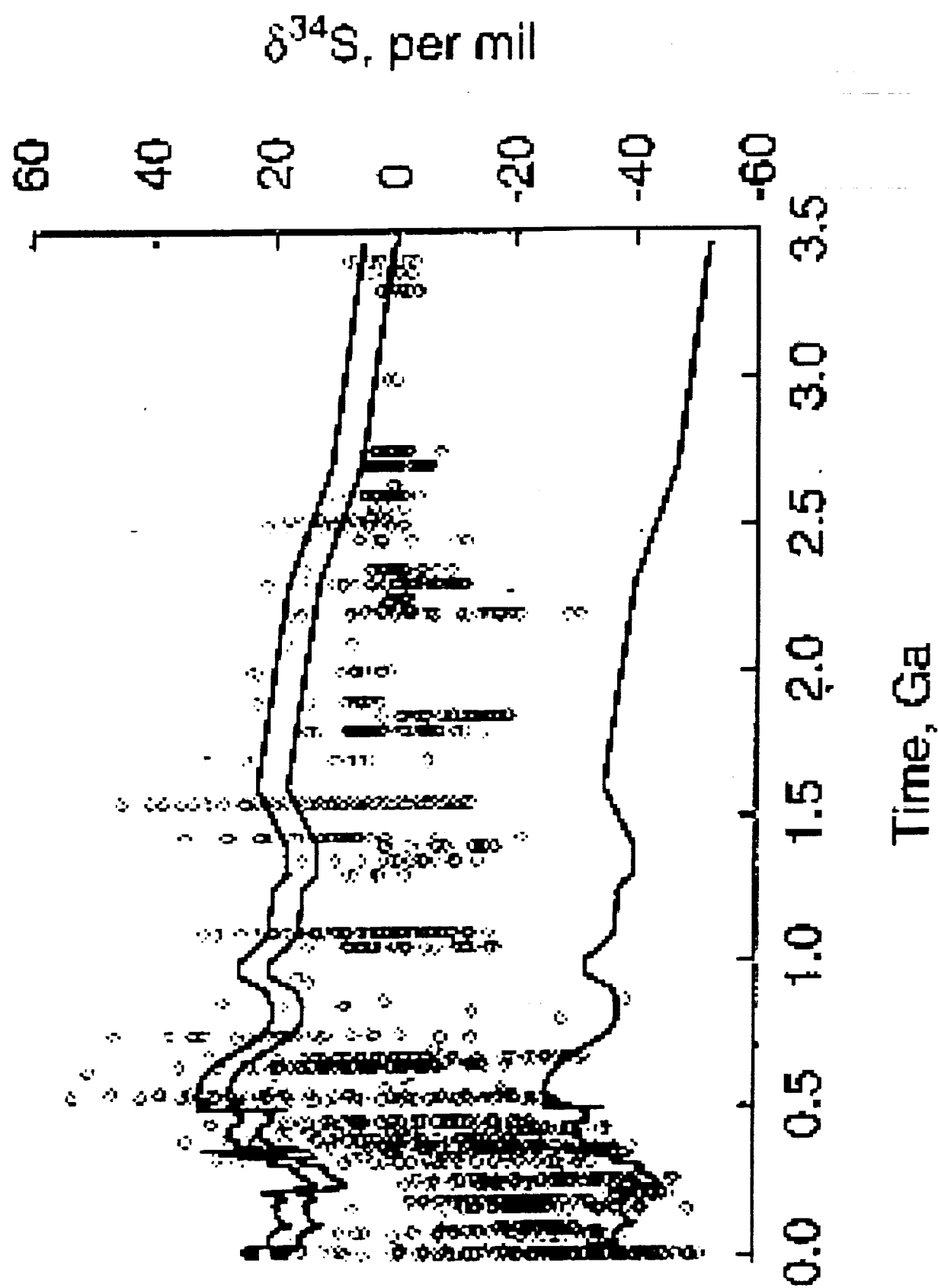


Isotopic Mass Balance of the Crustal Carbon Reservoirs



Carbon Isotopic Record in Sedimentary Carbonates and Organic Matter





II. Oxidizing

III. Aerobic

Atmosphere

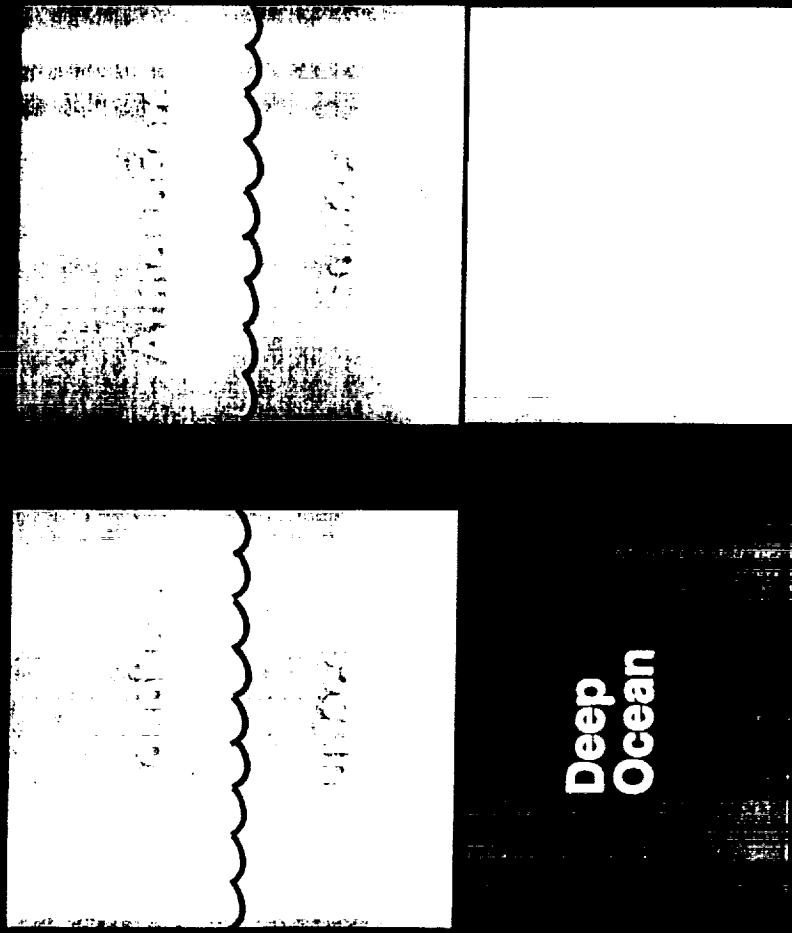
Surface
Ocean

Deep
Ocean

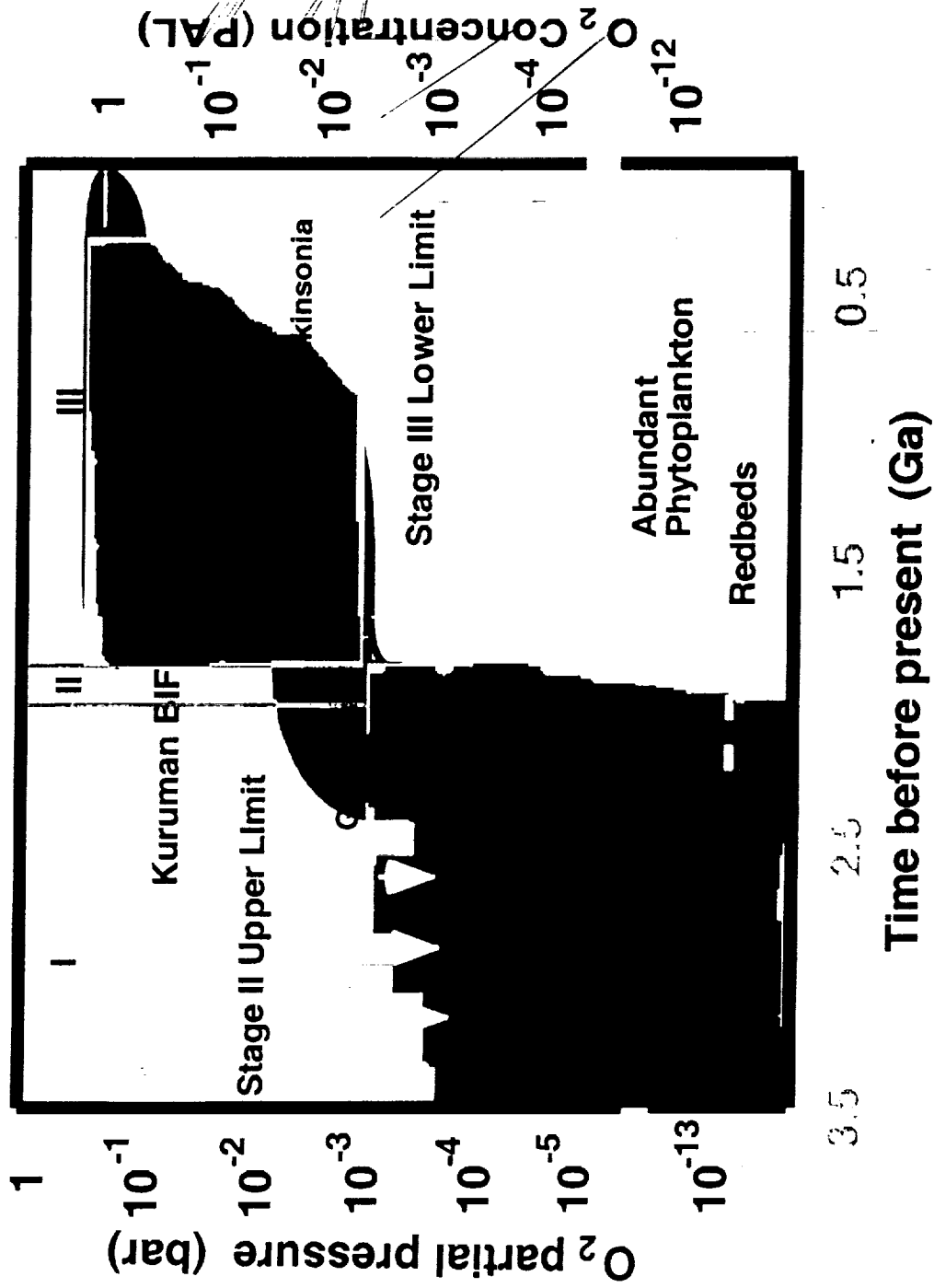
Deep
Ocean

Time

Kasting (1987)

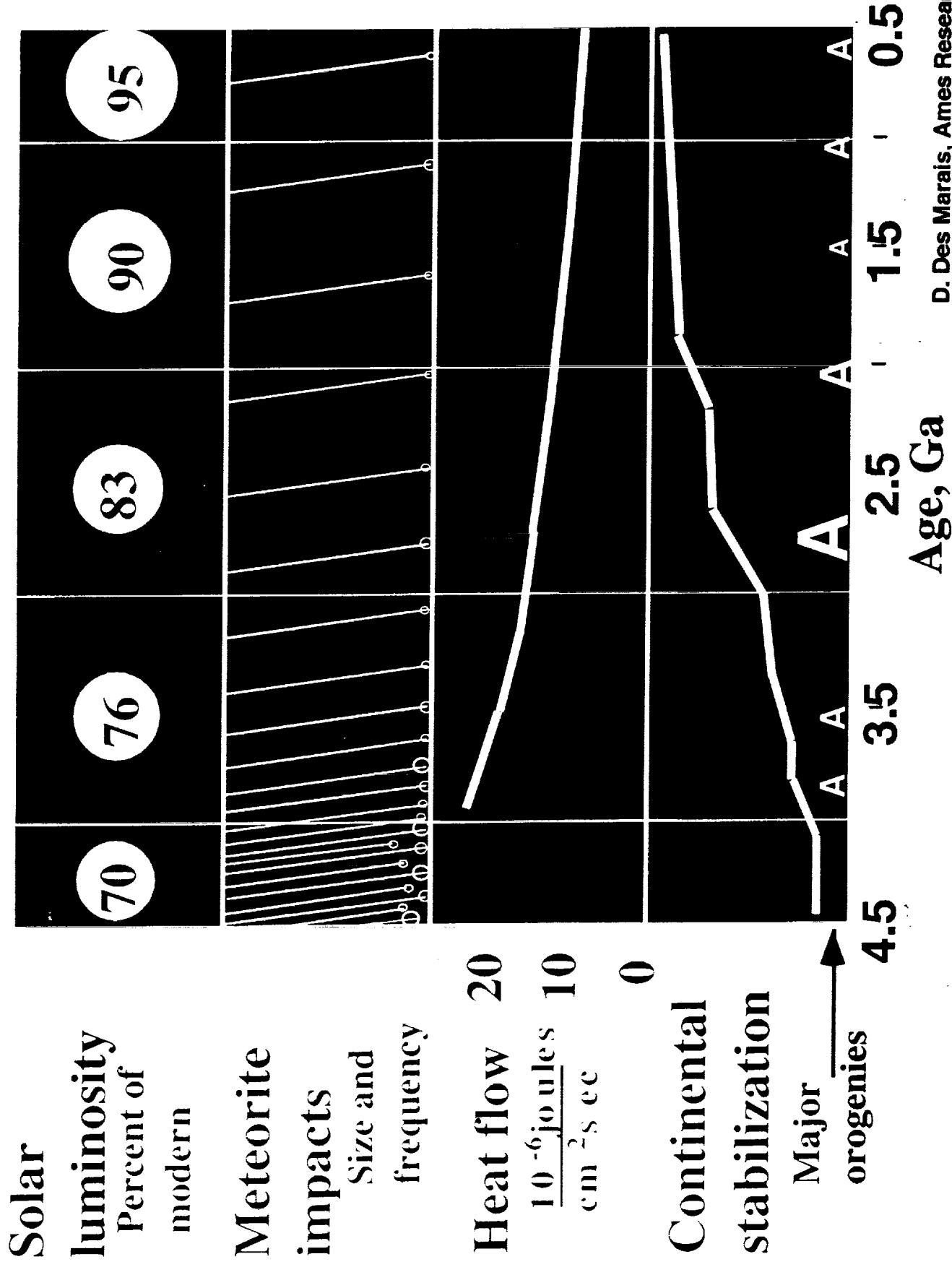


Rise of Atmospheric Oxygen

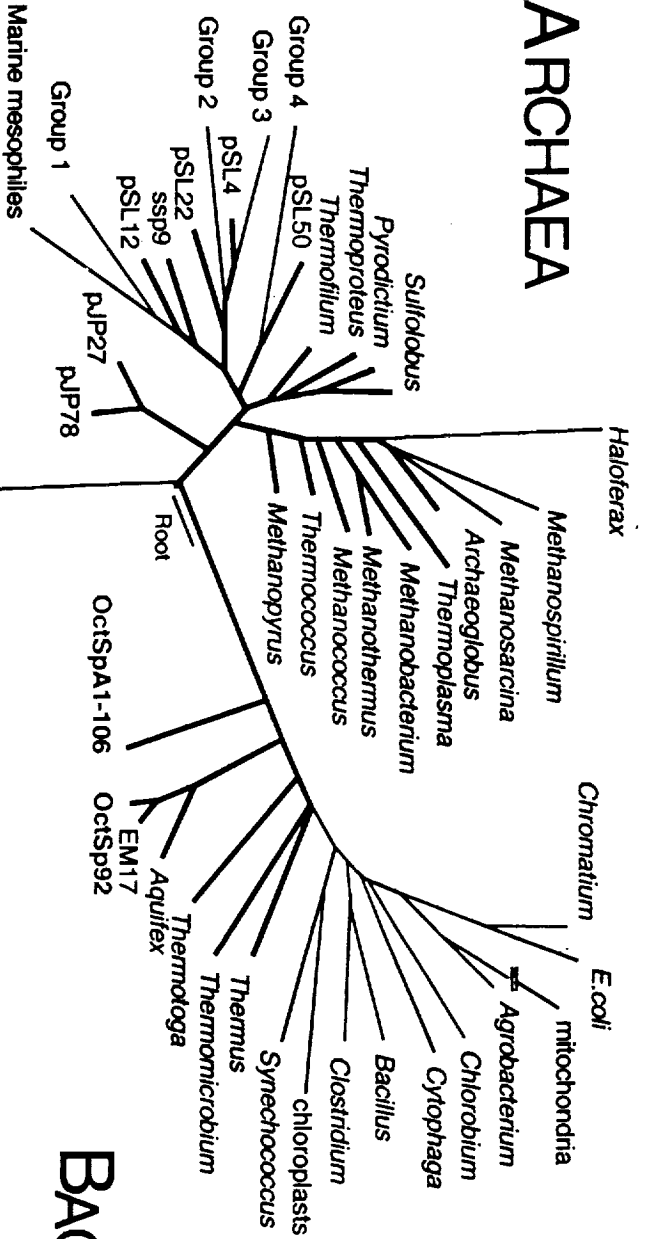


J. Kasting

Evolution of Earth's Early Environment

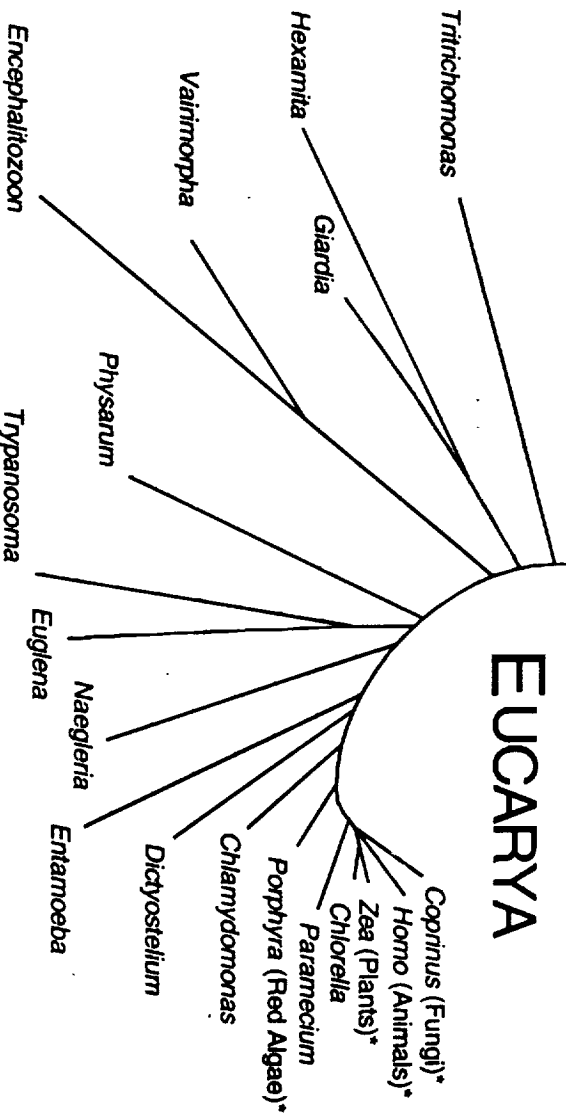


ARCHAEA

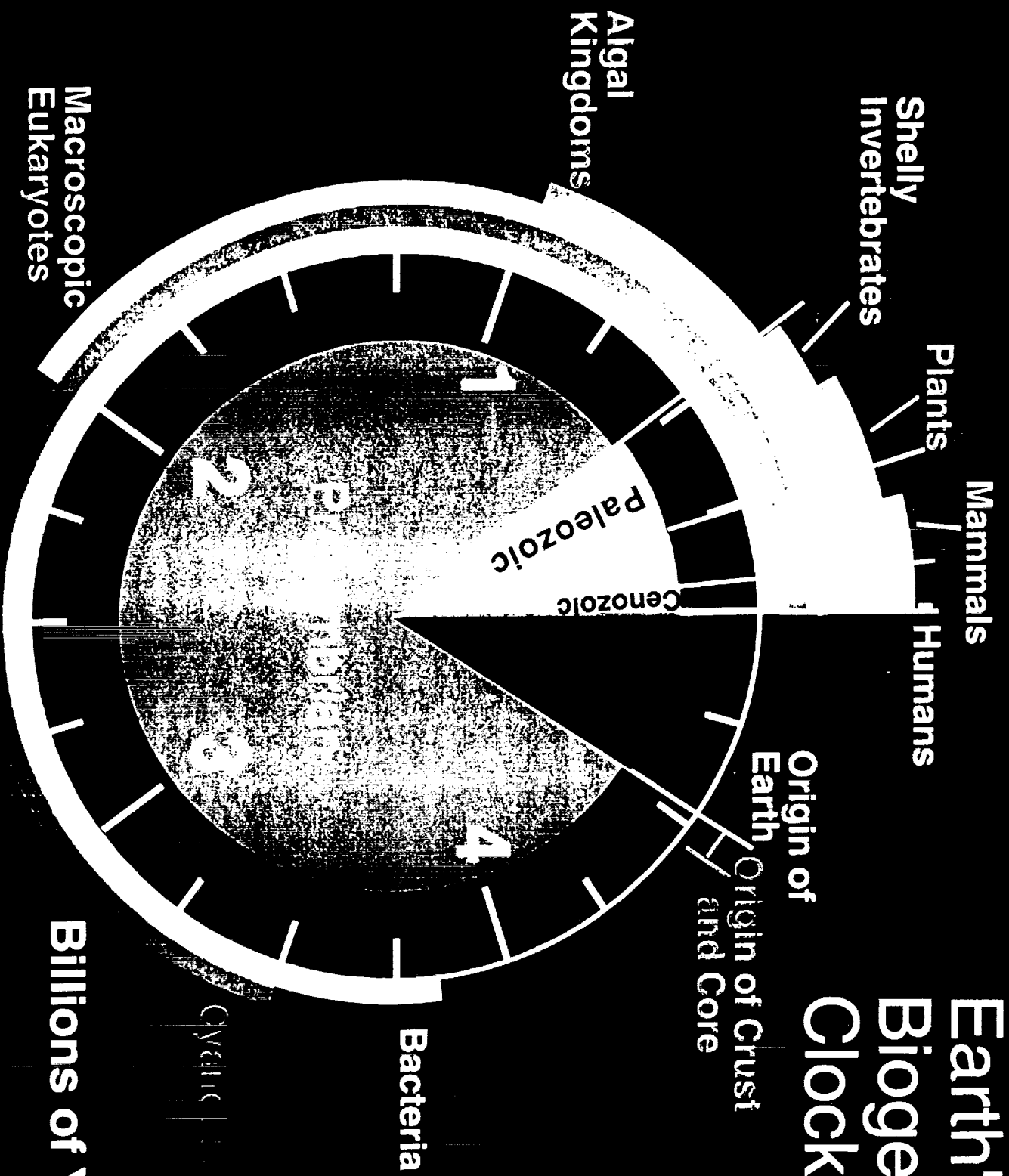


BACTERIA

EUCARYA



Earth's Biogeologic Clock



Billions of Years Ago

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NASA, Ames Research Center